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It can be dissolved in strong acid solutions to give the ionic complexes Sn(OH)2 3 2+ and Sn(OH) (OH)2 2 +, and in less acid solutions to give Sn 3 (OH) 4 2+. Note that anhydrous stannites, e.g. K 2 Sn 2 O 3, K 2 SnO 2 are also known. SnO is a reducing agent and is thought to reduce copper(I) to metallic clusters in the manufacture of so-called "copper ruby glass". Structure. Black, 7-SnO ...

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Cadmium sulfide is the inorganic compound with the formula CdS. Cadmium sulfide is a yellow solid. It occurs in nature with two different crystal structures as the rare minerals greenockite and hawleyite, but is more prevalent as an impurity substituent in the similarly structured zinc ores sphalerite and wurtzite, which are the major economic sources of cadmium.

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